Thermal Effects on Bioaccumulation and Antioxidant Enzyme Response in the Liver and Gills of Tilapia *Oreochromis niloticus* Exposed to Arsenic

EunYoung Min¹, Ji Won Jeong² and Ju-Chan Kang³*

¹Institute of Fisheries Science, Pukyong National University, Busan 619-911, Korea
²National Fishery Products Quality Management Service, Busan 600-016, Korea
³Department of Aquatic Life Medicine, Pukyong National University, Busan 608-737, Korea

Abstract

The effects of waterborne arsenic (As) exposure on bioaccumulation and antioxidant defenses were examined in the liver and gills of tilapia, *Oreochromis niloticus*, under thermal stress. Tilapia were exposed to different As concentrations (0, 200, and 400 μg L⁻¹) at three water temperatures (20, 25, and 30°C) for 20 days. After As exposure, higher levels of As accumulation were observed in the gills compared with the liver in elevated water temperatures. In terms of the antioxidant response, glutathione (GSH) levels and the activities of glutathione reductase (GR), glutathione peroxidase (GPx), and glutathione S-transferase (GST) significantly decreased in the liver and gills of tilapia exposed to As for 20 days, regardless of the As concentration (200 and 400 μg L⁻¹) or water temperature. These enzyme activities exhibited greater fluctuations in the liver and gills of tilapia after As exposure in water warmer than 20°C. The present findings suggest that the simultaneous stress of temperature change and As exposure can accelerate As accumulation and alter the antioxidant enzymes activities of tilapia.

Key words: *Oreochromis niloticus*, Arsenic, Water temperature, Bioaccumulation, Antioxidant enzymes

Introduction

Recently, Seong et al. (2010) reported that the sea surface temperatures of the East, West and South Seas of Korea have exhibited distinct increases. In aquaculture farming, water temperature is a key environmental factor affecting the growth and survival of aquatic organisms (Wang et al., 2006). In particular, the ability of fish to obtain oxygen from the environment can be strongly affected by water temperature; changes in temperature can alter the rate of oxygen consumption and its delivery to tissues at the physiological level (Powell and Watts, 2006). Previous reports have suggested that elevated temperatures can cause higher metabolic rates, which then lead to increased oxygen consumption and subsequent production of reactive oxygen species (ROS) (An and Choi, 2010). In aquatic environments, aquatic organisms are often exposed to multiple stressors, including temperature changes; therefore, understanding the influence of elevated temperature on the antioxidant profile of aquatic organisms is crucial (Bocchetti et al., 2004).

Due to adverse effects on human health, the contamination of aquatic ecosystems with arsenic (As) has been receiving worldwide attention, particularly in South and Southeast Asia (Kapaj et al., 2006). As is a potentially toxic trace element that tends to accumulate to a greater extent in aquatic compared with terrestrial animals (McGeachy and Dixon, 1989). The
United States Environmental Protection Agency set a limit of 0.01 ppm for As in drinking water (USEPA, 2001). Kim et al. (2007) reported that total As in freshwater and marine fish in Korea ranged from 0.024 to 2.23 and from 0.14 to 65.54 mg kg\(^{-1}\) dry wt, respectively. During a histopathological investigation, Ahmed et al. (2013) demonstrated that As is highly toxic to the liver and gills of tilapia, *Oreochromis mossambicus*, and that its toxicity increases with increasing As concentrations. Therefore, the present study used the gills and liver as vital organs for estimating the effects of As on bioaccumulation and antioxidant responses in tilapia.

As induces oxidative stress by cycling between metal oxidation states or by interacting with antioxidants (Halliwell, 2004). Liu et al. (2001) demonstrated that the free radical formation induced by As generates ROS that can interact directly with enzyme or metal-catalyzed processes in the liver of mice. As can exert its toxicity during a process of binding to critical thiol proteins and cellular non-protein thiol such as glutathione (GSH) (Chouchane and Snow, 2001). The regulation of the activity of antioxidant enzymes such as glutathione peroxidase (GPx), glutathione reductase (GR), and GSH-independent antioxidant by As is also an important aspect involved in the toxic mechanism of As (Shi et al., 2004). GSH-associated metabolism functions as a major mechanism for cellular protection against agents that generate oxidative stress. GSH participates in the detoxification process, scavenges free radicals, reduces peroxides, and conjugates with electrophilic compounds at several different levels (Dringen et al., 2000). This process is aided by enzymes such as GST, GPx, GR, aldo-keto reductase, and aldehyde dehydrogenase (Yu, 1994). Because antioxidant enzymes are widely used as biomarkers of marine pollution, research on antioxidant responses to marine environmental changes such as elevated water temperatures is tantamount. In the present study, we examined the activity of antioxidant enzymes after exposure to As at 20, 25, and 30°C in tilapia, *Oreochromis niloticus*. Our objective was to evaluate the combined effect of elevated temperature and As on bioaccumulation and the antioxidant profile (GSH, GST, GPx, and GR) in the liver and gills of *O. niloticus*.

### Materials and Methods

#### Experimental animals and water conditions

Tilapia *Oreochromis niloticus* (mass: 56.35 ± 8.77 g, length: 14.98 ± 1.78 cm; means ± SE), were sampled from a fish farm at Pukyong National University in Korea. After transport to the laboratory, healthy fish without visible damage were selected for 4 weeks of adaptation. Selected fish were maintained in 500 × 280 × 310-mm glass tanks.

#### Exposure protocol

Prior to temperature acclimation, fish were held for 4 weeks in aerated tanks in 20°C water to ensure a common thermal history between acclimation groups. Temperatures were then adjusted using a water heater (Electronic thermostat, MS701-H, Mink, Korea), starting at 20°C at increasing at 1°C intervals once every other day until the final target temperature was reached (25 or 30°C).

Two replications of toxicity tests (10 fish at each As concentration and each temperature) were conducted in static systems with continuous aeration at a constant water temperature (20, 25, or 30°C) and a constant room temperature of 20°C. Experimental fish were exposed to waterborne treatments of 0, 200, and 400 µg L\(^{-1}\) sodium arsenite (NaAsO\(_2\), Sigma-Alldrich, Inc., USA) for 20 days at each temperature. The tank water was changed thoroughly once every 2 days, at which time, water quality parameters were measured for the bioassay (Table 1).

#### Fish sampling

Fish exposed to each As concentration were sampled at the end of the experiment. Fish were anesthetized using benzocaine, and the standard length and weight were measured immediately. The liver and gills were removed and kept at -80°C until analysis.

#### Bioaccumulation

Each tissue was lyophilized to a constant weight after washing with distilled water and then digested using the wet digestion method after measuring the dry weight. Lyophilized organs were dissolved in HNO\(_3\) and re-dried by heating to 120°C. This process was repeated until all organic matter was digested completely. Final samples were dissolved in 2% HNO\(_3\) and used for analysis after filtration (Advantec MFS, Inc.; 0.2-µm filter). The concentration of total As in tissues was measured using an ICP-MS (Inductively Coupled Plasma Mass Spectrophotometer) (Elan, Perkin-Elmer Inc., USA) equipped with an automatic sampler using argon gas. An ICP multi-element standard solution VI (Merck) was used to produce a standard curve. As concentrations in the tissues of fish

### Table 1. Chemical components of seawater and experimental conditions used in this study

<table>
<thead>
<tr>
<th>Test parameters</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Culture type</td>
<td>Renewal 24 h toxicity test</td>
</tr>
<tr>
<td>Temperature (°C)</td>
<td>18.0 ± 0.5</td>
</tr>
<tr>
<td>pH</td>
<td>8.1 ± 0.5</td>
</tr>
<tr>
<td>Salinity (%)</td>
<td>33.5 ± 0.6</td>
</tr>
<tr>
<td>Dissolved oxygen (mg L(^{-1}))</td>
<td>7.1 ± 0.3</td>
</tr>
<tr>
<td>Chemical oxygen demand (mg L(^{-1}))</td>
<td>1.13 ± 0.1</td>
</tr>
<tr>
<td>Ammonia (mg L(^{-1}))</td>
<td>12.5 ± 0.7</td>
</tr>
<tr>
<td>Nitrate (mg L(^{-1}))</td>
<td>1.3 ± 0.3</td>
</tr>
<tr>
<td>Nitrate (mg L(^{-1}))</td>
<td>11.48 ± 1.0</td>
</tr>
</tbody>
</table>

http://dx.doi.org/10.5657/FAS.2015.0195
were expressed as μg g⁻¹ dry wt. The accumulation factor (AF) is often used to compare the body burden of an organism to the degree of contamination in the water. The following definition was used here:

$$AF = \frac{[\text{Me}]_{\text{exp}} - [\text{Me}]_{\text{control}}}{[\text{Me}]_{\text{water}}}$$

where $[\text{Me}]_{\text{exp}}$, $[\text{Me}]_{\text{control}}$, and $[\text{Me}]_{\text{water}}$ are the metal concentrations in the experimental group, control group, and water, respectively, measured in μg g⁻¹ (Holwerda, 1991).

Biochemical analysis

Liver and gill tissues used to determine enzyme activity were rinsed in 0.1 M KCl (pH 7.4) and homogenized (099CK4424, Glass-Col, Germany) in 0.1 M phosphate-buffered saline (pH 7.4). The homogenate was centrifuged at 10,000 g for 60 min (+4°C), and the supernatant was used for the GR, GPx, and GST assays. To determine GSH, tissues were homogenized in 20 mM Tris buffer (pH 7.8) containing 0.25 M sucrose. The homogenate was centrifuged at 10,000 g for 30 min (+4°C), and the supernatant was used for assays. The supernatant was stored at -75°C prior to analysis. Protein concentrations were determined using the method of Bradford (1976), with bovine serum albumin as a standard.

Glutathione (GSH)

The GSH level was measured following the method of Srivastava and Beutler (1970). A precipitation solution (3 µL) containing metaphosphoric acid, disodium ethylenediamine tetraacetic acid (Na₂EDTA), and NaCl was added to a 2-mL sample. The mixture was centrifuged at 4,500 g for 5 min. The supernatant (2 mL) was added to 8 mL of a 0.3 M Na₂HPO₄ solution and 1 mL 0.5 mM 5,5-dithiobis-2-nitrobenzoic acid (DTNB) in that order. The GSH level was evaluated at 412 nm and determined using the reduced GSH standard curve. The GSH level was expressed as nmol GSH mg⁻¹ protein.

Glutathione reductase (GR)

GR activity was determined using NADPH (nicotinamide adenine dinucleotide phosphate) and the GSH oxidized (GSSG) as substrate, as described by Beutler (1984). One unit of GR was defined as the NADPH consumed per minute while catalyzing the reduction of 1 mM GSSG.

Glutathione peroxidase (GPx)

GPx activity was measured by determining the decrease in NADPD (Sigma, USA) during the formation of reduced GSH via GR at 340 nm using a modified method of Bell et al. (1985). In this procedure, 2.5 mM H₂O₂ in 50 mM potassium phosphate buffer (1 mM GSH, 0.1 M NADPH, 0.5 U/mL GR, 1 mM EDTA, 2 mM sodium azide, pH 7.4) was used as the substrate.

Glutathione S-transferase (GST)

GST activity was measured using a modified method of Habig (1974). The sample (20 µL) was mixed with 100 µL 0.2 M potassium phosphate (pH 6.5) and 40 µL distilled water. The reaction was initiated by adding 20 µL 10 mM GSH and 20 µL 10 mM CDNB (1-chloro-2,4-dinitrobenzene). The reaction mixture was kept at room temperature for 1 min. GST activity was then determined by the absorbance increase at 340 nm after 5 min and was expressed as nmol min⁻¹ mg⁻¹ protein.

Data analysis

Data are presented as means ± standard error. For statistical analysis, a one-way analysis of variance (ANOVA) was used, followed by Duncan’s multiple range tests using SPSS statistical software (SPSS Inc., Chicago, IL, USA). Differences were considered statistically significant when $P < 0.05$.

Results

Bioaccumulation

Fig. 1 presents the levels of As accumulation in the liver and gills of tilapia, O. niloticus, at different water temperatures and As exposure concentrations over 20 days. The accumulation of As in tilapia resulted in a net increase in the total As content in tissues compared with the non-exposed (control) group. As concentrations ranged from 0.29 to 0.3 µg g⁻¹ dry wt in the liver and from 0.45 to 0.55 µg g⁻¹ dry wt in the gills in the non-exposed group. In the liver and gills, there was a significant increase in As accumulation at 30°C compared with 20°C (Fig. 1). Although the metal contents in the liver were similar at 20°C (0.48 ± 0.07 µg/g) and 25°C (0.57 ± 0.08 µg/g), the liver showed the highest level of As accumulation (0.71 ± 0.11 µg/g) at 400 µg/L exposure at 30°C. As accumulation in the liver and gills of tilapia increased with the water temperature, and this pattern did not differ with the exposure level of As ($P > 0.05$).

The AFs for liver and gills at 200 and 400 μg L⁻¹ of waterborne As exposure at different water temperatures are presented in Table 2. The AFs increased with the water temperature in the liver and gills; however, an inverse relationship was observed between AFs and As exposure concentration.

Biochemical analysis

Glutathione (GSH)

Fig. 2 presents GSH levels in the liver and gills of tilapia
In the liver compared with the non-exposed group at each temperature evaluated \((P < 0.05)\), GSH levels did not differ between 200 and 400 \(\mu\text{g L}^{-1}\) exposure \((P > 0.05)\). At 25 and 30°C, the GSH levels in the gills significantly decreased after As treatment but not at 20°C. Glutathione reducatase (GR)

The GR activity measured in the liver of tilapia increased significantly at 30°C compared with both 20 and 25°C \((P < 0.05)\). After 20 days of As exposure, there was a significant decrease in GR activity in the liver compared with the non-exposed group at each water temperature. In particular, liver GR activity decreased significantly with increased water temperature. In the gills, GR activity decreased significantly after As treatment, regardless of the water temperature, but values did not significantly differ between 200 and 400 \(\mu\text{g L}^{-1}\).
As treatment at each water temperature evaluated, although warmer water temperatures (25 and 30°C) caused a decrease in GST activity compared with values at 20°C in the non-exposed group.

Discussion

Metal accumulation in fish tissues depends on the metal concentration, exposure time, manner of metal uptake, environmental conditions (e.g., water temperature, pH, hardness, salinity), as well as intrinsic factors such as fish age and feeding habits (Campana et al., 2003). Although various metals exhibit different affinities to fish tissues, most accumulate mainly in the liver, kidney, and gills. The accumulation of metals in various organs of fish may cause structural lesions and functional disturbances (Jezierska and Witeska, 2006).
the present study, waterborne As exposure led to similar As levels between the liver and gills of tilapia (Fig. 1). Takatsu et al. (1999) also reported similar levels of As accumulation between the liver and gills of the freshwater fish, Tribolodon hakonensis. As accumulation in the liver affects how this organ functions in detoxification, while accumulation in the gills affects its role in respiration, excretion, and osmoregulation (Culioli et al., 2009). Furthermore, Culioli et al. (2009) have suggested that the gills accumulate the highest levels of As and represent the best organ for biomonitoring As pollution.

In the present study, As levels in tilapia exhibited higher accumulation when water temperatures exceeded 20°C (Fig. 1). Increased accumulation of metals at higher temperatures likely results from higher rates of metal uptake through increased respiration and ventilation in the gills. Kock et al. (1996) reported that elevated cadmium and lead levels in Salvelinus alpinus indicated higher uptake rates of both metals in the summer, during which metabolic rate increased with increases in water temperature. Pereira et al. (2010) suggested that higher temperatures promote the accumulation of Mn, Cr, Pb, and Cd, particularly in the most burdened organs of the feral fish Liza aurata.

Currently, oxidative stress is the most widely accepted and well-studied mechanism of As toxicity (Ercal et al., 2001). Previous studies have also reported As-induced antioxidant imbalances (Samuel et al., 2005; Valko et al., 2006; Hughes and Kitchin, 2006). An antioxidant response is the primary cellular defense mechanism against As toxicity. The metabolism of As may follow several potential pathways, i.e., classical reduction, oxidative methylation, and GSH conjugation. As-induced oxidative stress can cause GSH to react directly with electrophiles; therefore, GSH may play an important role in As detoxification along with the cofactors GPx and GST (Dopp et al., 2010).

As induces a depletion of antioxidant defense enzymes such as SOD, GPx, GR, and GST, although the pattern of depletion varies with factors such as dose, time, and organ (Mishra et al., 2008). Generally, short-term exposure to low As concentrations increases the activities of these enzymes, whereas chronic exposure usually decreases activity levels (Shi et al., 2004). Lantz et al. (2007) proposed that GST could be used as a biomarker for As toxicity. Ventura et al. (2007) reported that GST activities were reduced in the polychaete Laeonereis acuta after 7 days of exposure to 500 μg As L⁻¹. Likewise, our study demonstrated a significant decrease in antioxidant enzymes such as GPx, GR, and GST in tilapia exposed to low As concentrations (200 and 400 μg L⁻¹) for 20 days. Moreover, this effect of As was particularly pronounced at elevated water temperatures. Specifically, in the non-exposed group, GSH and GR activities were significantly higher at 30°C compared with the lower temperature treatments. However, As exposure induced a decline in these enzymes at 30°C compared with lower temperatures (Figs. 2 and 3). The reduction in GR activity was accompanied by a reduction in GSH level; thus, the inhibition of GR activity by As may reduce the availability of GSH for the maintenance of cellular redox status (Thomas et al., 2001).

Bajanyukova et al. (2007) reported that GPx and GST appear to play particularly important roles in the antioxidant defenses of goldfish subjected to thermal stress (from 3 to 23°C).

Given that As is enzymatically methylated in the liver, its toxicity depends on the type of As compound and the species of fish, as well as environmental factors such as water temperature (Kotsanis and Iliopoulos, 1999). In addition, Cherkasov et al. (2006) demonstrated that Cd-induced oxidative stress in oysters can be particularly strong and pronounced at an elevated water temperature of 30°C compared with 20°C. These authors suggested that antioxidants offer adequate protection against Cd-induced ROS at 20°C in oysters; in contrast, at 30°C, antioxidant systems appear to be unavailable for defense against Cd-induced ROS generation, and considerable oxidative damage occurs.

**Fig. 5.** Glutathione S-transferase (GST) activities in liver (A) and gill (B) tissues tilapia *O. niloticus* exposed to various arsenic concentrations at 20, 25 and 30°C for twenty-days. Each point represents a mean value ± S.D of three replicates. Vertical bar denotes a standard error (n = 10). Values with different superscript are significantly different (P < 0.05) as determined by Duncan’s multiple range test.
Kaur et al. (2005) reported that high-temperature treatments (from 20 to 32°C) modulated antioxidant enzymes such as GSH, GR, and GST in the liver and gills of Channa punctata. They suggested that although the pattern of increase or decrease did not follow a uniform pattern in every tissue type, the alterations were quite unambiguous. Similarly, in the present study, these same enzymes in the liver did not exhibit the same patterns as those manifested in the gills under the simultaneous stress of temperature change and As exposure. Likewise, the activities of antioxidant enzymes did not exhibit the same pattern in various tissues of common carp, Cyprinus carpio, exposed to As (Ventura et al., 2009). These authors suggested that antioxidant responses to As depend upon the tissue type, which may be a strategy specific to each tissue to protect cells from As toxicity. Likewise, fluctuations in antioxidant enzymes such as GSH, GPx, and GST declined, indicating that homeostasis (i.e., a protective measure of organisms to oxidative stress) ceases through a failure to induce compensatory measures or antioxidant enzymes due to As exposure.

Antioxidant defense is one of the most efficient mechanisms of protecting biological molecules, but the process is also a major target for toxic compounds. Altered levels of these antioxidants provide insight into the degree, extent, or stage of As-induced oxidative damage. Thus, antioxidant levels can be used as biomarkers of As toxicity. In addition, the antioxidant activity of enzymes such as GSH, GPx, and GST is directly and inversely proportional to As toxicity, depending on water temperature. Furthermore, the inhibition or fluctuation of antioxidant enzymes indicates that the fish might not possess a normal physiological homeostasis against oxidative stress due to As exposure. These findings support our results, which demonstrated that the exposure of tilapia to As caused significant modulations in antioxidant enzymes at an elevated temperature of 30°C. In particular, the present study suggests that the large fluctuations in antioxidant enzymes at 30°C are due to accelerated As accumulation in tilapia caused by elevated water temperatures. Similarly, Carvalho and Fernandes (2005) reported that the toxicity of heavy metals in fish increased more at 30°C compared with 20°C due to increased oxygen consumption and respiration rate.

Our data indicate that As affected antioxidant responses in the liver and gills of tilapia in terms of GSH levels and GPx, GR, and GST activities. The antioxidant defenses were likely overwhelmed in the liver and gills, leading to increased As accumulation and exaggerated As toxicity when the water temperature was increased. Furthermore, our study provides another mechanism for the synergistic effects observed between environmental temperature and pollutants, which may have important implications for the survival of fish in polluted environments during seasonal warming and/or global climate change.

Min et al. (2015) As Bioaccumulation and Antioxidant Responses in Tilapia Oreochromis niloticus

References


Yu BP. 1994. Cellular defenses against damage from reactive oxygen species. Physiol Rev 74, 139-162.